

CARBON-14 IN LUNAR SOIL, E.L. Fireman, J. DeFelice, and J. D'Amico,
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Measurements of the carbon-14 temperature release from lunar surface soils give measurable amounts of carbon-14 released between 600 C and 1000°C; however, no carbon-14 was released at these temperatures for subsurface soils (1,2,3). We assigned a solar-wind origin to the 600-1000°C component of carbon-14. If so, this carbon-14 component in surface soils should exhibit the same type of inverse grain-size dependence (i. e., surface correlation) as the rare gases (4,5). Des Marais *et al.* (6) observed an inverse grain-size dependence for the carbon concentration for size fractions of < 100 μM. We therefore undertook a study of the grain-size dependence of the 600-1000°C released carbon-14.

We were allocated ~1.0-g splits of the < 10, 10-30, 30-37, 37-74, 74-125 μM fractions of 10084 soil separated by King *et al.* (7). King *et al.* (7) sieved 14.63 g of 10084 soil and obtained 1.34 g of the < 10 μM, 3.74 g of the 10-30 μM, 1.31 g of the 30-37 μM, 2.51 g of the 37-74 μM, 1.86 g of the 74-125 μM, and 3.83 g of the > 125 μM sized grains.

Our procedures for carbon extraction, conversion of the extracted compounds to CO₂, and CO₂ purification have not changed (1). We extract the carbon compounds from a lunar soil sample in a molybdenum crucible, by raising its temperature with resistance heating and then holding its temperature constant for 4 hr while the gases are removed with an automatic toeppler pump. The temperatures were 400, 600, 800, and 1000 °C. We heat the sample above its melting point for 4 hr by induction heating and remove the gases. We then repeat the melting at a higher (>1500°C) temperature and remove the gases again. The collected gas is passed over CuO at 650°C, which converts the carbon compounds to CO₂. The CO₂ is condensed in a trap at -196°C, recovered from the trap at -78°C, and measured in a standard volume. To purify the CO₂ from radon, the CO₂ is converted to CO over Zn at 300°C. The gases that condense from CO at -196°C are removed. The purified CO is then reconverted to CO₂ over CuO. On the basis of stoichiometry, no carbon is lost in the Zn purification.

The amounts of CO₂ obtained from an extraction ranged between 0.11 and 2.1 cm³ STP. The CO₂ is put into a small proportional counter, the CO₂ pressure is kept under 1 atm and argon is added to the counter until the pressure of the gas is 1.5 atm. In our previous measurements (1,2), only counters of 0.7 cm³ volume were used, and, on occasion, we had to count aliquots. We have constructed similar counters of 2.5 cm³ volume and the more appropriately sized counter is now used for counting total amount of CO₂. Counters of both sizes have energy resolutions of 20%. The counter backgrounds were determined with similar gas fillings using CO₂ from a petroleum source. The 0.7 and 2.5 cm³ counters had backgrounds of approximately 3 and 6 counts/day, respectively, above 7 keV energy. Carbon-14 activity is distinguished most easily from the counter background above 7 keV. When filled with less than 0.5 atm of CO₂, the efficiencies for counting carbon-14 above 7 keV are 25 and 35 ± 3% for the smaller and larger counters, respectively. The backgrounds and efficiencies rise with CO₂ pressure above 0.5 atm. When the CO₂ pressure is above 1.0 atm, the operating voltages approach 2500 V and breakdown pulses occur.

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Table 1 gives the 400-1000°C temperature-released carbon-14 from the 10084 samples that we analyzed. There was little carbon-14 released at 400°C from any of the size fractions. At 600°C, there was significant carbon-14 released from the <10 μM fraction but not from the larger fractions. At 800°C, the <10 μM fraction released 26.5 dpm/kg of carbon-14, which was three times the amount released from the 10-30 μM and six times the amount released from the 30-37 μM fractions. However, at 1000°C, the <10 μM fraction released about the same amount of carbon-14 as the 30-37 μM fraction and less than that released from the 10-30 μM fraction. The sum of the 600-1000°C releases are 40.4 ± 2.5 , 26.0 ± 2.5 , 7.8 ± 1.7 , and 3.7 ± 1.7 dpm/kg of carbon-14 for the <10, 10-30, 30-37, and 74-125 μM fractions, respectively. Within the accuracy of the measurements, the 600-1000°C carbon-14 is inversely proportional to the mean grain diameters.

The carbon-14 in the melts that followed the 1000°C heatings and the remelts are also given in Table 1. The amount of carbon-14 in the melt of the <10 μM fraction, 36.0 ± 2.7 dpm/kg, was similar to the amount obtained from the 74-125 μM fraction, 30.1 ± 5.0 dpm/kg. There were smaller amounts of carbon-14 from the melts of 10-30 and 30-37 μM fractions; however, on the basis of remelts, not all the carbon-14 was extracted from the 30-37 μM fraction. Reedy and Arnold (8) calculated the spallation production rates for carbon-14 from galactic cosmic rays to be 19 dpm/kg. Begemann et al. (9) measured the total carbon-14 in bulk 10084 soil to be 39 ± 4 dpm/kg. The total carbon-14 in the <10 μM fraction is twice the bulk value and four times the value calculated for galactic cosmic rays.

References

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Table 1. Carbon-14 temperature-release patterns.

Sample	10084.941	10084.1519	10084.939	10084.937
Weight (g)	1.26	1.01	1.28	1.00
Grain size (μ)	<10	10-30	30-37	74-125
Temp.	^{14}C	^{14}C	^{14}C	^{14}C
$^{\circ}\text{C}$	(dpm/kg)	(dpm/kg)	(dpm/kg)	(dpm/kg)
400	2.1 ± 0.9	2.3 ± 1.4	0 ± 0.8	2.2 ± 1.2
600	8.9 ± 0.9	0 ± 1.1	0.5 ± 0.8	2.4 ± 1.2
800	26.5 ± 2.0	8.0 ± 1.2	4.2 ± 1.3	1.5 ± 1.2
1000	<u>5.0 ± 0.8</u>	<u>18.0 ± 1.8</u>	<u>3.6 ± 1.0</u>	<u>-0.2 ± 1.2</u>
Sum (600-1000)	40.4 ± 2.5	26.0 ± 2.5	7.8 ± 1.7	3.7 ± 2.1
Melt	36.0 ± 2.7	21.3 ± 2.1	4.0 ± 1.5	30.1 ± 5.0
Remelt	5.8 ± 2.3	3.6 ± 2.1	11.0 ± 1.0	-3.0 ± 2.5