

10062
Ilmenite Basalt (low K)
78.5 grams



Figure 1: Closeup photo of 10062,13 showing vugs in ilmenite basalt. This is the largest remaining piece (25g). NASA#S76-21515. Scale in mm.

Introduction

10062 is an old, low-K, high-Ti basalt from Apollo 11. Based on texture and composition, James and Wright (1972) termed 10062 an ophitic ilmenite basalt, whereas Gamble et al. (1978) describe the texture as subophitic (figure 1). At 3.8 b.y., basalt 10062 is one of the oldest lunar mare basalts (other than those found as clasts in breccias). It has been exposed to cosmic rays for 90 m.y. (the apparent age of West Crater).

Petrography

Schmitt et al. (1970) termed 10062 as a “coarse-grained, vuggy, ophitic, olivine basalt.” Carter and MacGregor (1970), Beaty and Albee (1978), Gamble et al. (1978) and Kramer et al. (1977) provide petrographic descriptions of 10062. The texture is ophitic, with radiating lath-like and acicular crystals

of plagioclase, intergrown with tabular and skeletal crystals of ilmenite, irregular grains of zoned clinopyroxene and olivine (figure 2). Late stage silica, fayalite, ulvöspinel, troilite and apatite are present in the mesostasis.

The abundance of ilmenite is striking, with some grains as large as ~ 1 mm. Some ilmenite has relict cores of armalcolite, which appears to have been replaced and overgrown by ilmenite.

The origin of high-Ti basalts has been discussed by many authors (see for example Walker et al. 1975). Since 10062 is about the same age as 10003 and 10029 (Papanastassiou et al. 1977), and has similar mineralogy, chemistry and texture, Beaty and Albee



Figure 2: Photomicrograph of thin section of 10062 showing ophitic texture. Field of view about 2 mm. NASA S76-26268.

(1978) and Gamble et al. (1978) have concluded these three samples may be related to each other.

Mineralogy

Olivine: Carter and MacGregor (1970) and Beatty and Albee (1978) report olivine compositions ranging Fo₇₆₋₅₀. Rounded grains of olivine are often included in clinopyroxene suggesting a reaction relationship.

Pyroxene: Hollister and Hargraves (1970) describe the compositional zoning of pyroxene in 10062. Pigeonite appears to be exsolved on a fine scale. Pyroxene compositions were given by Beatty and Albee (1978) (figure 3).

Plagioclase: The composition of plagioclase is An₉₄₋₈₀ (Beatty and Albee 1978).

Ilmenite: Ilmenite crystals in 10062 are lath-shaped, often skeletal, and up to 1600 microns in long dimension (Gamble et al. 1978). Carter and MacGregor (1970) found lamellae of rutile and blebs of spinel exsolved from the larger grains of ilmenite.

Armalcolite: LSPET (1969) and Carter and MacGregor (1970) reported rounded grains of dull grey armalcolite in the cores of ilmenite grains.

Mineralogical Mode of 10062

	Carter and MacGregor 1970	Gamble et al. 1978	Beatty and Albee 1978
Olivine	5 vol. %	4.3	5.0
Pyroxene	52	42.8	39
Plagioclase	24	33.7	37.6
Ilmenite	18	14.2	12.5
Cristobalite	tr.	2.4	4.6
Mesostasis	1	1.9	0.6

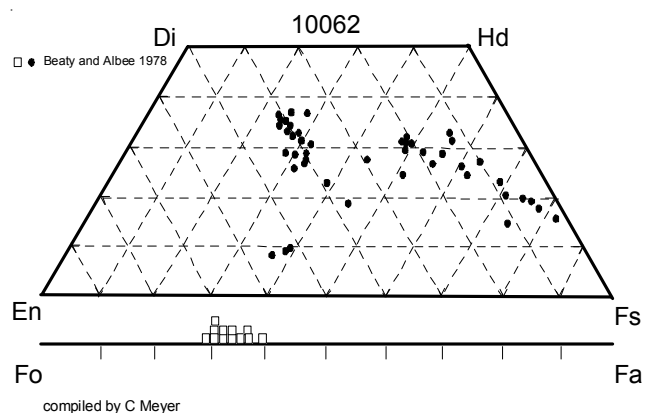


Figure 3: Olivine and pyroxene composition for 10062 (data replotted from Beaty and Albee 1977).

Silica: Cristobalite is present in abundance (5%) in 10062.

Chemistry

10062 is a high-Ti, low-K Apollo 11 basalt (figure 4). The chemical composition of 10062 is tabulated in table 1. The rare earth elements (REE) and large ion lithophile elements (LIL) are depleted with respect to the high-K, Apollo 11 basalts (see section on 10057). There is a hint that La and Ce are also slightly depleted with respect to the other REE (figure 5).

Radiogenic age dating

Turner (1970) and Guggisberg et al. (1979) dated 10062 by the Ar-Ar plateau method as 3.82 b.y. and 3.84 b.y. (respectively). Papanastassiou et al. (1977) dated 10062 by Rb-Sr and Sm-Nd isochron methods and determined an even older age (figures 6 and 7). Other low-K, high-Ti Apollo 11 basalts are slightly younger and may be from a different flow(s).

Cosmogenic isotopes and exposure ages

Turner et al. (1970) reported an exposure age of 90 m.y. by ^{38}Ar .

Other Studies

O'Hara et al. (1974) experimentally determined the liquid composition of 10062 (at 1120 deg. C) cotectic with olivine, clinopyroxene, anorthite and ilmenite at low pressure and made the case for the hypothesis that mare basalts are slowly consolidated giant lava lakes and not the result of partial melting of the deep lunar interior! However, Walker et al. (1975) review the origin of titaniferous lunar basalts by partial melting

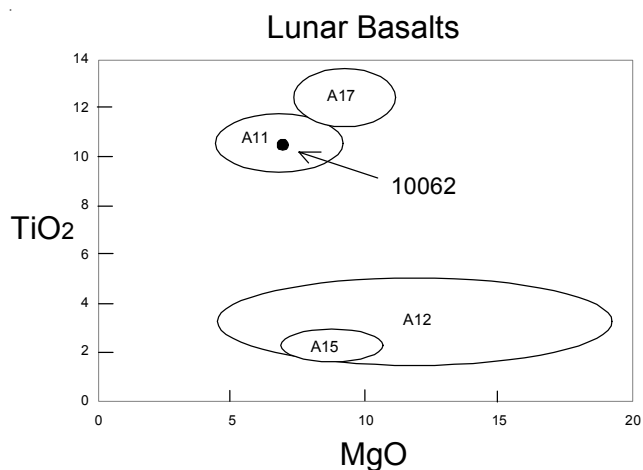


Figure 4: Lunar basalt sample 10062 is a typical high Ti basalt.

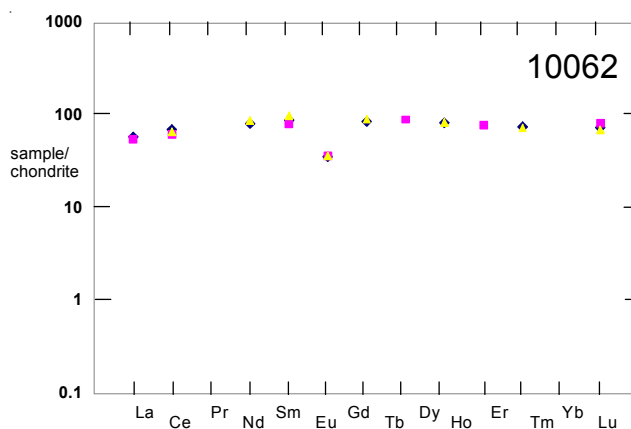


Figure 5: Normalized rare-earth-element diagram for 10062 (data from Gast et al. 1970, Goles et al. 1970, and Philpotts et al. 1970). Note the slight depletion of La and Ce.

and conclude that the preferred source region is a “late-stage ilmenite-rich cumulate produced from the residual liquid of the primordial differentiation of the outer portion of the Moon.”

The concentrations of Sm, Nd, Lu and Hf and the isotopic ratios of $^{143}\text{Nd}/^{144}\text{Nd}$ and $^{176}\text{Hf}/^{177}\text{Hf}$ were determined by Unruh et al. (1984).

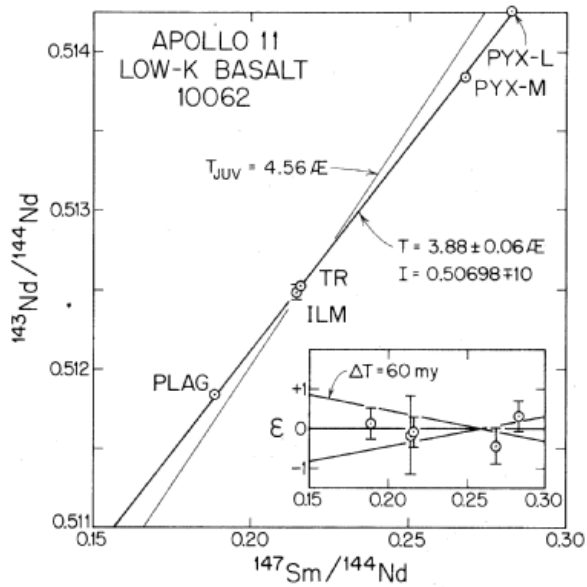


Figure 6: Sm-Nd internal mineral isochron of mare basalt 10062 (from Papanastassiou et al. 1977).

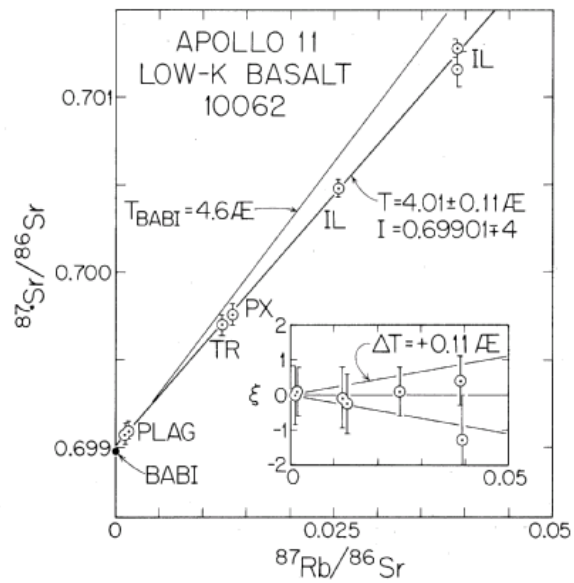


Figure 7: Rb-Sr internal mineral isochron of mare basalt 10062 (from Papanastassiou et al. 1977).

Processing

There are 13 thin sections.

List of Photo #s for 10062

- S69-46511 – 521 PET B&W
- S69-49219
- S69-49226 – 230
- S69-49237 – 238
- S76-26268 TS
- S76-27271 TS
- S76-21515 – 518 ,13 ,14
- S76-22210 – 211 ,33
- S70-49023 – 026 TS
- S70-50019 – 020 TS

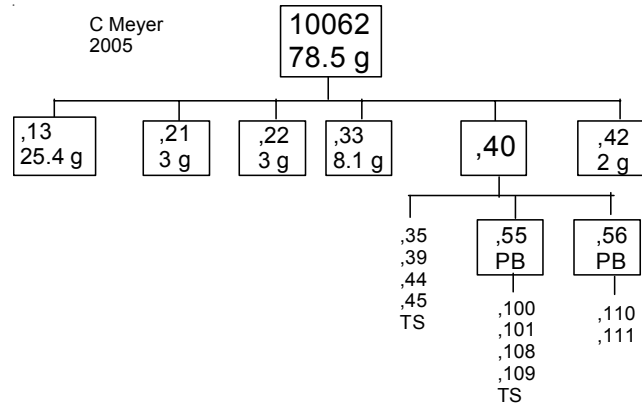


Table 2

	U ppm	Th ppm	K ppm	Rb ppm	Sr ppm	Nd ppm	Sm ppm	technique
Compston et al. 1970		0.9		0.89	187.8			idms
Gast and Hubbard 1970			628	0.9	141.6	37	13.3	idms
Philpotts and Schnetzler 1970			601	0.832	194	39.9	14.7	idms

Summary of Age Data for 10062

	Ar-Ar plateau	Rb-Sr	Sm-Nd
Turner et al. 1970	3.82 ± 0.06 b.y.		
Papanastassiou et al. 1977		4.01 ± 0.11	3.88 ± 0.06
Guggisberg et al. 1979	3.84 ± 0.04		

Caution: Data not corrected for new decay constants.

Table 1a. Chemical composition of 10062.

reference weight	Compston70	Gast 70	Rose 70	Goles 70	Philpotts70	Turekian70	Kharkar 71	Gast70
SiO ₂ %	39.8	(a)	38.8	(a)				
TiO ₂	10.74	(a) 10.7	(b) 10.3	(a)			11.5	(c)
Al ₂ O ₃	10.22	(a)	12.1	(a)				
FeO	19.22	(a)	18.3	(a)		16.7	18.01	(c)
MnO	0.3	(a)	0.27	(a) 0.23	(c)	0.19	0.26	(c)
MgO	7.08	(a)	7.21	(a)				
CaO	11.47	(a) 10.8	(b) 12	(a)			11.6	(c)
Na ₂ O	0.41	(a) 0.44	0.69	(a) 0.42	(c)	0.4	0.4	(c) 0.44
K ₂ O	0.08	(a) 0.076	(b) 0.07	(a)	0.07	(b) 0.03		(c) 0.076 0.08 (b)
P ₂ O ₅	0.12	(a)						
S %	0.16	(a)						
sum	99.6							
Sc ppm				74.7	(c)	76	86	(c)
V				75	(c)			
Cr			1710	(a) 1540	(c)	1410	1660	(c)
Co				13.8	(c)	13	13	(c)
Ni		13	(b)			15		(c)
Cu						4		(c)
Zn		11	(b)					
Ga	3	(a)						
Ge ppb								
As								
Se						0.23		(c)
Rb	0.89	(a) 0.9	(b)		0.832	(b)		0.9 0.81 (b)
Sr	187.8	(a) 141.6	(b)		194	(b)		142 196 (b)
Y	103	(a)						
Zr	319	(a)		290	(c)			
Nb								
Mo						0.16		(c)
Ru								
Rh								
Pd ppb								
Ag ppb						80		(c)
Cd ppb								
In ppb								
Sn ppb								
Sb ppb								
Te ppb								
Cs ppm								0.032 (b)
Ba		79.9	(b)	230	(c) 134	(b)		80.5 140 (b)
La		13.8	(b)	13.1	(c)	12	11.5	(c) 14.2 14.5 (b)
Ce		42.7	(b)	38	(c) 40.2	(b) 48	37.6	(c) 42.7 44.8 (b)
Pr								
Nd		37	(b)		39.9	(b)		36.7 37.5 (b)
Sm		13.3	(b)	11.9	(c) 14.7	(b) 10	8.7	(c) 13.3 13.7 (b)
Eu		2.02	(b)	2.07	(c) 2.07	(b) 1.8	2.2	(c) 2.02 2.06 (b)
Gd		17.2	(b)		18.1	(b)		17.7 18.2 (b)
Tb				3.3	(c)			
Dy		20.4	(b)		20.6	(b) 22	24.6	(c) 20 20.4 (b)
Ho				4.4	(c)			
Er		12	(b)		11.8	(b)		12.4 12.8 (b)
Tm								
Yb		12.1	(b)	13.5	(c) 11.3	(b) 6.3	7.8	(c) 12.1 12.3 (b)
Lu		1.73	(b)	1.94	(c) 1.76	(b) 0.87	1.7	(c) 1.73 (b)
Hf				11.8	(c)	10	11.9	(c)
Ta				1	(c)	1.8	1.7	(c)
W ppb								
Re ppb								
Os ppb								
Ir ppb								
Pt ppb								
Au ppb						5.8		(c)
Th ppm	0.9	(a)						
U ppm				0.27	(c)	0.28		(c)

technique (a) XRF, (b) IDMS, (c) INAA

Table 1b. Chemical composition of 10062.

reference	Rhodes 80	Wiesmann75		
weight	1.25 g	166 mg	199mg	
SiO ₂ %	40.21	(a)		
TiO ₂	10.34	(a)		
Al ₂ O ₃	10.82	(a)		
FeO	19.08	(a)		
MnO	0.28	(a)		
MgO	7.01	(a)		
CaO	11.57	(a)		
Na ₂ O	0.43	(c)	0.44	
K ₂ O	0.08	(a)	0.076	0.08 (b)
P ₂ O ₅	0.14	(a)		
S %				
sum				
Sc ppm	80	(c)		
V				
Cr	1750	(c)		
Co	13.1	(a)		
Ni				
Cu				
Zn				
Ga	7.8	(a)		
Ge ppb				
As				
Se				
Rb	1.2	(a)	0.9	0.81 (b)
Sr	202	(a)		196 (b)
Y	107	(a)		
Zr				
Nb				
Mo				
Ru				
Rh				
Pd ppb				
Ag ppb				
Cd ppb				
In ppb				
Sn ppb				
Sb ppb				
Te ppb				
Cs ppm			0.032	(b)
Ba				140 (b)
La	13.5	(c)	14.2	14.5 (b)
Ce	46	(c)	42.7	44.8 (b)
Pr				
Nd			36.7	37.5 (b)
Sm	13.2	(c)	13.3	13.7 (b)
Eu	2	(c)	2.02	2.06 (b)
Gd			17.7	18.2 (b)
Tb	2.8	(c)		
Dy			20	20.4 (b)
Ho				
Er			12.4	12.8 (b)
Tm				
Yb	10.7	(c)	10.8	10.9 (b)
Lu	1.66	(c)	1.73	(b)
Hf	10.6	(c)		
Ta	1.8	(c)		
W ppb				
Re ppb				
Os ppb				
Ir ppb				
Pt ppb				
Au ppb				
Th ppm	0.8	(c)		
U ppm				
technique	(a) XRF, (b) IDMS, (c) INAA			

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