

10085
Coarse-fines
569 grams

DRAFT



Figure 1: Selected coarse fines from 10085. Scale is in mm. Photo from Wood et al. 1969.

Introduction

10085 and 10084 were created during the Apollo 11 preliminary examination by sieving a large portion of 10002. 10084 was what passed through a 1 mm sieve and 10085 was what did not (figure 1). However, it is not likely that sieving such a large mass would have gone to completion, so 10085 would still have had a lot of < 1 mm material.

The coarse-fine particles from 10085 were widely distributed during the first allocations of lunar samples (e.g. Chao et al. 1970, Wood et al. 1970, King et al. 1970, von Englehardt et al. 1970). An interesting aspect of lunar samples is that they are fine-grained enough for a sample on the order of 100 mg, or 2 mm in size, to generally be large enough to represent a “whole rock”.

Petrography

There are ~250 thin sections of 10085 material. This is how the first anorthosites were recognized (Wood et al. 1970) and Lunny Rock # 1 was probably found this way (Albee and Chodos 1970). A large metal particle (88 mg), termed “minimoon”, was found in 10085 (figure 2).



Figure 2: “Minimoon”. 3.5 mm. Goldstein et al. 1970.

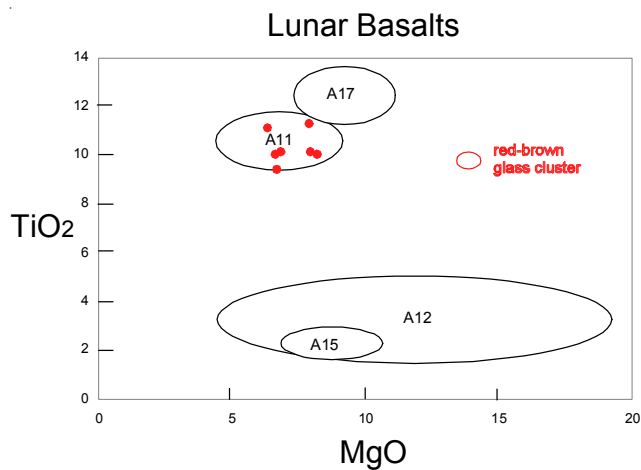


Figure 3: Composition of small basalt fragments and of tightly-clustered volcanic glass in coarse-fines.

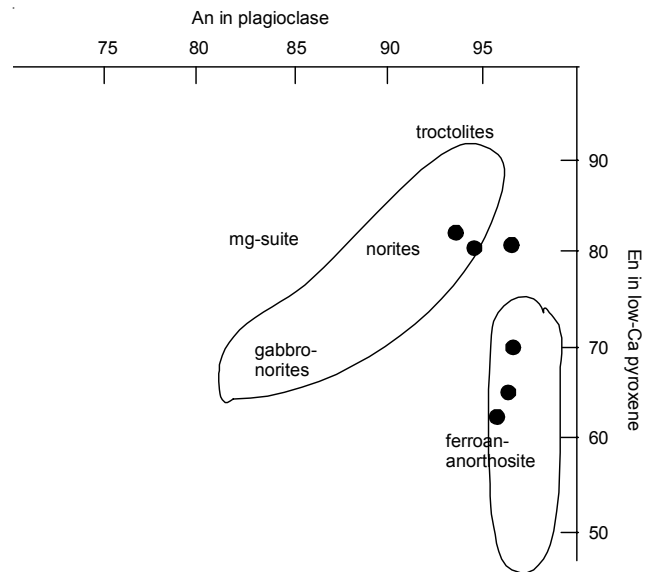


Figure 4: Composition of pyroxene and plagioclase in some of the white rocks from 10085 (from Simon et al. 1983).

King et al. (1970) state “ sample 10085,11 contained 1227 grains as follows: mafic holocrystalline rock fragments, 585; microbreccia, 395; glass splatter and agglomerates, 204; regularly shaped glass (spheres, dumbbells etc), 6 ; anorthosite and other light colored rock fragments, 37.” Wood et al. (1970) reported 37% basalt, 52% soil breccia, 4% glass, 2% anorthosite, 1.5% anorthosite glass and 1.5% anorthositic breccia.

Beaty et al. (1979) and Grove and Beaty (1980) studied numerous small basalt samples (figure 3, table 1).

Simon et al. (1983) and Laul et al. (1983) studied the small white particles in 10085 (figure 4, table 2).

Chemistry

Laul et al. (1983) analyzed 38 small white rock fragments from 10085 (table 2) and Beaty et al. (1979) analyzed a number of small basalt fragments (table 1).

Processing

The sieving of 10084, 10085 is a mystery – see section on 10084. In the 1980s, Simon picked white particles out of ,104, followed by Larry Taylor who picked out basalt particles – see processing photos from data pack.

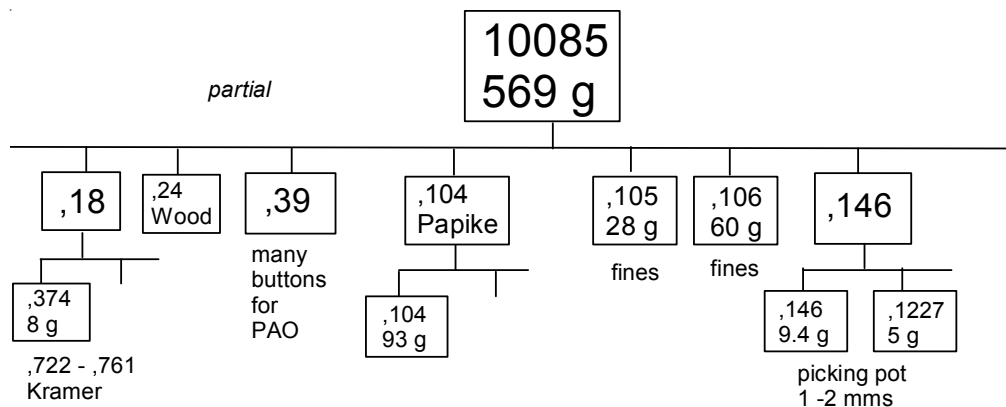
Table 1. Chemical composition of 10085 particles (basalts).

reference	Beaty 79								
weight									
SiO ₂ %									
TiO ₂	10.2	11.6	10.2	10.1	9.3	10.3	12.1	11.1	(a)
Al ₂ O ₃	8	8.2	8.1	8.1	8.7	8.2	8.8	8.3	(a)
FeO	20	21.5	20.2	20.4	20	20.6	20.4	19.5	(a)
MnO	0.25	0.25	0.25	0.25	0.25	0.24	0.23	0.25	(a)
MgO	8	8	8	7	7	7	8	7	(a)
CaO	10	11.1	10.6	11	10.8	10.1	10.7	10	(a)
Na ₂ O	0.5	0.5	0.52	0.51	0.58	0.49	0.5	0.48	(a)
K ₂ O	0.26	0.25	0.32	0.32	0.34	0.31	0.28	0.3	(a)
P ₂ O ₅									
S %									
sum									
Sc ppm	84	89	82	84	76	84	85	87	(a)
V	66	84	63	66	52	56	79	65	(a)
Cr									
Co	26	29	27	27	24	26	27	28	(a)
Ni									
Cu									
Zn									
Ga									
Ge ppb									
As									
Se									
Rb									
Sr									
Y									
Zr	520	360	430	440	400	500		370	
Nb									
Mo									
Ru									
Rh									
Pd ppb									
Ag ppb									
Cd ppb									
In ppb									
Sn ppb									
Sb ppb									
Te ppb									
Cs ppm									
Ba	230	240	330	320	380	330	290	400	
La	23	22.3	27.2	26.9	33	28.7	25.2	25.7	
Ce	75	72	83	80	98	100	90	79	
Pr									
Nd	62	58	71	70	80	74	64	65	
Sm	19.2	18.2	21.6	21.1	24.6	22.6	20.1	20.6	
Eu	2.11	2.06	2.25	2.26	2.55	2.55	2.19	2.18	
Gd									
Tb	4.4	4.2	4.7	4.6	5.2	5	4.6	4.5	
Dy	31	29						30	
Ho									
Er									
Tm									
Yb	15.5	15.1	17.3	17.2	19.6	18.2	16.3	16.8	
Lu	2.27	2.24	2.55	2.54	2.86	2.63	2.32	2.46	
Hf	14.4	15.3	15.9	14.9	16.7	15.8	15.1	15.4	
Ta	2.5	2.2	2.6	2.3	2.6	2.4	2.2	2.7	
W ppb									
Re ppb									
Os ppb									
Ir ppb									
Pt ppb									
Au ppb									
Th ppm	1.8	1.6	2.1	1.9	2.8	2.3	2	2.1	
U ppm									
technique:	(a) INAA								

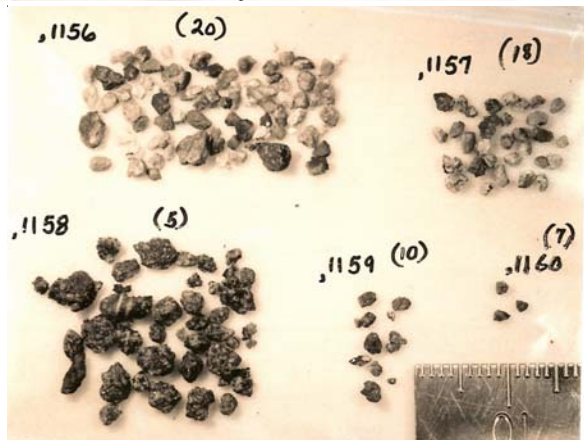
Table 2. Chemical composition of 10085 particles (white rx).

reference weight	(see ref for more data)													
	Laul 83		granulitic				anor.		gabbro			norite		
SiO2 %	Poik.													(a)
TiO2	0.4	0.56	0.5	0.3	0.29	0.82	0.2	0.2	0.1	0.3	0.3	0.37	0.3	(a)
Al2O3	29	19.9	22.8	25.5	30	27.8	32	35.5	34	20.6	27	25.5	16.5	(a)
FeO	2.7	7	7	5.4	2.7	4.2	1	0.17	0.3	7	7.9	5.55	8	(a)
MnO														
MgO	2	10	9	7.5	4	5	0.5	0.5	0.5	11.5	2	6.4	17.5	(a)
CaO	16.9	13.5	14.3	16	16.2	16.2	18.9	19.6	18.5	12.9	16.5	15.8	9.3	(a)
Na2O	0.54	0.48	0.34	0.36	0.57	0.56	0.37	0.29	0.42	0.26	0.35	0.34	0.3	(a)
K2O	0.14	0.05	0.061	0.055	0.04	0.048	0.009	0.009	0.01	0.015	0.02	0.052	0.03	(a)
P2O5														
S %														
sum														
Sc ppm	6	21	16.3	10.5	5.1	5.4	2.3	0.52	0.48	10	13	15	14	(a)
V	10	30	50	30	20	20				20	20	40	110	(a)
Cr														
Co	6	12.6	50	18	6.6	9.9	2.1	0.75	0.7	15	6	19	40	(a)
Ni	40	20	300	150	20	30				20		200		(a)
Cu														
Zn														
Ga														
Ge ppb														
As														
Se														
Rb														
Sr	180	130	150	160	230	200	170		160	140	160	170	110	
Y														
Zr														
Nb														
Mo														
Ru														
Rh														
Pd ppb														
Ag ppb														
Cd ppb														
In ppb														
Sn ppb														
Sb ppb														
Te ppb														
Cs ppm														
Ba	90	40	50	60	30	50						80	90	
La	5.5	2.6	2.5	3.7	2.3	2.5	0.42	0.28	0.15	1.25	0.68	3.6	8	
Ce	13	8.5	6.1	8.7	5.9	5.7				3	2	8.8	20	
Pr														
Nd	8.4	9	4	5.4	4	4				2.3		6.5	12	
Sm	2.4	3.1	1.2	1.7	1	1	0.21	0.08	0.06	0.7	0.35	2	3.3	
Eu	1.2	1	0.82	0.85	1.3	1.3	0.83	0.8	0.89	0.71	0.85	0.87	1.1	
Gd														
Tb	0.5	0.88	0.3	0.4	0.21	0.23				0.16		0.46	0.8	
Dy	3.3	8	2.1	2.5	1.5	1.4	0.29			1.2	0.65	3.3	5.5	
Ho														
Er														
Tm														
Yb	2	4.6	1.55	1.55	0.9	1	0.2			0.73	0.45	2.1	4	
Lu	0.29	0.71	0.22	0.21	0.12	0.12	0.032		0.01	0.11	0.07	0.31	0.81	
Hf	1.6	2.5	1.4	1.1	0.8	0.95	0.21			0.35		0.97	1.1	
Ta	0.24		0.23	0.15	0.1	0.19				0.08		0.08	0.16	
W ppb														
Re ppb														
Os ppb														
Ir ppb														
Pt ppb														
Au ppb														
Th ppm	0.9			0.6	0.3	0.14				0.12		0.5	0.82	
U ppm	0.3			1										

technique: (a) INAA



Processing photos for ,104 and ,1161.



References for 10085.

Agrell S.O., Scoon J.H., Muir I.D., Long J.V.P., McConnell J.D.C. and Peckett A. (1970) Observations on the chemistry, mineralogy and petrology of some Apollo 11 lunar samples. *Proc. Apollo 11 Lunar Sci. Conf.* 93-128.

Albee A.L. and Chodos A.A. (1970) Microprobe investigations on Apollo 11 samples. *Proc. Apollo 11 Lunar Science Conf.* 135-157.

Albee, Burnett, Chodos, Eugster, Huneke, Papanastassiou, Podosek, Price, Sanz, Tera and Wasserburg G.J. (1970) Ages, irradiation history, and chemical composition of lunar rocks from the Sea. *Science* **167**, 463-466.

Beaty D.W., Hill S.M.R., Albee A.L., Ma M.-S., and Schmitt R.A. (1979a) The petrology and chemistry of basaltic fragments from the Apollo 11 soil, part 1. *Proc. 10th Lunar Sci. Conf.* 41-75.

Brown G.M., Emeleus C.H., Holland J.G. and Phillips R. (1970) Mineralogical, chemical, and petrological features of Apollo 11 rocks and their relationship to igneous processes. *Proc. Apollo 11 Lunar Sci. Conf.* 195-219.

Chao E.C.T., James O.B., Minkin J.A., Boreman J.A., Jackson E.D. and Raleigh C.B. (1970) Petrology of unshocked crystalline rocks and evidence of impact metamorphism in Apollo 11 returned lunar samples. *Proc. Apollo 11 Lunar Sci. Conf.* 287-314.

Goldstein J.I., Henderson E.P. and Yakowitz H. (1970) Investigation of lunar metal particles. *Proc. Apollo 11 Lunar Sci. Conf.* 499-512.

Grove T.L. and Beaty D.W. (1980) Classification, experimental petrology and possible volcanic histories of the Apollo 11 high-K basalts. *Proc. 11th Lunar Planet. Sci. Conf.* 149-177.

King E.A., Carman M.F. and Butler J.C. (1970) Mineralogy and petrology of coarse particulate material from the lunar surface at Tranquillity base. *Proc. Apollo 11 Lunar Sci. Conf.* 599-606.

King E.A. *and a cast of thousands* (1969) Lunar Sample Information Catalog, Apollo 11. Lunar Receiving Laboratory, MSC 412 pp

Kramer F.E., Twedell D.B. and Walton W.J.A. (1977) Apollo 11 Lunar Sample Information Catalogue (revised). Curator's Office, JSC 12522

Laul J.C., Papike J.J., Simon S.B. and Shearer C.K. (1983) Chemistry of the Apollo 11 highland component. *Proc. 14th Lunar Planet. Sci. Conf.* B139-149. *JGR* **88**

LSPET (1969a) Preliminary examination of lunar samples. In Apollo 11 Prelim. Sci. Rpt. NASA SP-214. 123-142

LSPET (1969b) Preliminary examination of lunar samples from Apollo 11. *Science* **165**, 1211-1227.

Simon S.B., Papike J.J., Shearer C.K. and Laul J.C. (1983) Petrology of the Apollo 11 highland component. *Proc. 14th Lunar Planet. Sci. Conf. in J. Geophys. Res.* **88**, B103-138.

von Engelhardt W., Arndt J., Muller W.F. and Stoffler D. (1970) Shock metamorphism of lunar rocks and origin of the regolith at the Apollo 11 landing site. *Proc. Apollo 11 Lunar Sci. Conf.* 363-384.

Wood J.A., Marvin U.B., Powell B.N. and Dickey J.S. (1970) Mineralogy and petrology of the Apollo 11 lunar sample. *Smithson. Astrophys. Observ. Spec. Rep.* 307

Wood J.A., Dickey J.S., Marvin U.B. and Powell B.N. (1970a) Lunar anorthosites. *Science* **167**, 602-604.

Wood J.A., Dickey J.S., Marvin U.B. and Powell B.N. (1970b) Lunar anorthosites and a geophysical model of the Moon. *Proc. Apollo 11 Lunar Sci. Conf.* 965-988.