DISCOVERY OF ANDRADITE GARNET AND EVIDENCE FOR HIGH TEMPERATURE HYDROTHERMAL PROCESSES (> 300 °C) IN THE LOWER YAXCOPOIL-1 IMPACT-MELT BRECCIAS. H. E. Newsom1, T. Salge2, M. J. Nelson1, and M. N. Spilde1, MSC03 2050, 1Univ. of New Mexico, Institute of Meteoritics, Dept. of Earth and Planetary Sciences, Albuquerque, NM 87131 U.S.A. Email: newsom@unm.edu. 2Bruker AXS Microanalysis GmbH, Schwarzschildstrasse 12, 12489 Berlin, Germany.

Introduction: Hydrothermal processes in large impact craters on Mars may have provided important environments for life and these processes can be studied in analog impact craters such as the 180 km diameter Chicxulub impact [1-3]. The Yaxcopoil-1 drill core in the annular trough of the Chicxulub crater exhibits layered impactites consisting of suevite and impact melt breccias (Fig. 1)[4-7]. The use of element mapping techniques at the thin section scale (Fig. 1) reveals a complex sequence of mineral growth and dissolution, coupled with the emplacement of matrix material and alteration due to hydrothermal processes.

Evidence for hydrothermal processes in large craters on Mars may have provided important environments for life and these processes can be studied in analog impact craters such as the 180 km diameter Chicxulub impact [1-3]. The Yaxcopoil-1 drill core in the annular trough of the Chicxulub crater exhibits layered impactites consisting of suevite and impact melt breccias (Fig. 1)[4-7]. The use of element mapping techniques at the thin section scale (Fig. 1) reveals a complex sequence of mineral growth and dissolution, coupled with the emplacement of matrix material and alteration due to hydrothermal processes.

Results: Microprobe and XRD analyses were conducted at the University of New Mexico, USA, and additional SEM work was conducted by Dr. Tobias Salge in Berlin. The XRD results show the presence of smectite clays in the matrix material, but no evidence for chlorite or other high temperature hydrothermal minerals. However, element mapping reveals the presence of additional evidence for both the high and low temperature hydrothermal systems. Zoned andradite garnet with Fe-rich rims and Ti-rich cores was identified in the matrix based on morphology and chemical composition [8] (Table 1). The garnets are partially dissolved (probably by lower temperature fluids, as seen in the inset to Fig. 2. Other common minerals present in the matrix include calcite, halite, and potassium-feldspar grains, which are found in the matrix as isolated grains and in the rims of the Al-rich clasts. Element mapping of the lower melt breccia units at the Yaxcopoil-1 drill core also provide evidence for multiple stages of fracturing of the Al-rich melt clasts, K-metasomatism and injection of Mg-rich material with slightly different chemical signatures.

Discussion: The andradite garnet supports the presence of an early high temperature hydrothermal system. Natural occurrences suggest temperatures > 350 °C [9]. The garnet composition also reflects changes in any variable that affect the aqueous speciation of Fe and Al, notably temperature, fO2, pH and the salinity of the hydrothermal solution, and changes in the Al/Fe ratio of the pore fluid at the site of crystal
The properties of the garnet, especially the negative values of the exchange component TiMg[Fe\textsuperscript{3+}]-\textsubscript{2}, are consistent with hydrothermal rather than igneous or metamorphic garnets (e.g., Russell et al [11]), reflecting a higher fO\textsubscript{2} in the fluid. Furthermore, the exchange components suggest, based on the zoning profile, that during formation of the garnets the fluids increased in SiO\textsubscript{2} activity and increased in fO\textsubscript{2}, possibly reflecting penetration of surface water.

The later evolution of the hydrothermal system apparently led to partial dissolution of the garnets due to lower temperatures. Interestingly, hydrothermal garnets (partially resorbed) were also found in the central uplift of the 36 km diameter Manson structure by McCarrville and Crossey [9], although these garnets are almost void of Ti, and may represent formation under lower fO\textsubscript{2} conditions based on the exchange coefficients.

The nature of the Mg-rich matrix material is still in doubt and must reconcile evidence for multiple stages of emplacement, an extensive hydrothermal alteration epsode and a chemical composition that could be consistent with a silic-dolomitic melt.

**Table 1** Preliminary EDS analyses (normalized to 100%) of garnet in the upper right of Fig. 2. The TiO\textsubscript{2} variation due to zoning is reflected in the center versus rim abundances.

<table>
<thead>
<tr>
<th></th>
<th>CaO</th>
<th>FeO</th>
<th>SiO\textsubscript{2}</th>
<th>Al\textsubscript{2}O\textsubscript{3}</th>
<th>TiO\textsubscript{2}</th>
<th>MgO</th>
<th>Total</th>
</tr>
</thead>
<tbody>
<tr>
<td>mean</td>
<td>36.1</td>
<td>22.6</td>
<td>31.7</td>
<td>4.6</td>
<td>4.1</td>
<td>0.8</td>
<td>100</td>
</tr>
<tr>
<td>centre</td>
<td>34.7</td>
<td>22.5</td>
<td>31.1</td>
<td>4.6</td>
<td>6.2</td>
<td>1.0</td>
<td>100</td>
</tr>
<tr>
<td>rim</td>
<td>35.4</td>
<td>24.9</td>
<td>32.2</td>
<td>4.0</td>
<td>3.0</td>
<td>0.5</td>
<td>100</td>
</tr>
</tbody>
</table>

**Conclusions:** The following sequence of events formed the lower portion of the Yax-1 breccias: (1) Impure dolomite and silicate basement lithologies were melted and ejected during crater formation, (e.g Stöffler et al [12]), (2) The silicate melt was quenched, brecciated, and enriched in potassium by seawater, or another K-rich fluid during transport or shortly after deposition, and (3) This deposit was later permeated by Mg-rich silicic-dolomitic melt, possibly from melt bodies in the ejecta deposit that were not as well mixed as seen in the upper ejecta material. Alternatively, the early/high temperature hydrothermal system could have drastically altered a composition similar to the Al-rich melt clasts. This hydrothermal system at a temperature >300 °C resulted in early K-metasomatism, formation of zoned andradite garnets, and alteration of the Mg-rich matrix material to smectite clay. The high temperature is consistent with a diopside-hedenbergite, scapolite, sphenic and apatite assemblage identified by Zürcher and Kring [7]. Later lower temperature fluids deposited halite and resulted in some late clay deposition [4]. The late stages of the hydrothermal system may have introduced the Li, Be and B measured with the ion probe in the matrix by Newsom et al. [13].


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