

ABUNDANCES OF C, N, H, He, AND S IN APOLLO 17 SOILS FROM STATIONS 3 AND 4: IMPLICATIONS FOR SOLAR WIND EXPOSURE AGES AND REGOLITH EVOLUTION.

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Abundances of C, N, S, H, He, and metallic Fe have been measured in two suites of Apollo 17 fines. One suite of samples is a trench sequence from the rim of Ballet Crater at station 3. These include: 73221,7, a skim sample of the thin 0.5 cm layer of medium gray soil; 73241,5, a light gray soil from the upper 5 cm of the trench which overlies a marbled zone; and 73261,8 and 73281,6, which represent the medium gray and light gray portions of the marbled zone, respectively. The second suite of samples includes 74220,84, the orange soil, and its companion gray soils 74240,16 and 74260,7. The methods of gas extraction by acid hydrolysis and stepwise vacuum pyrolysis followed by combustion are described elsewhere (Sakai et al.).

Consistent intra-sample abundance patterns for solar wind derived C, N, and H species show up clearly in the data in Tables 1 and 2. These patterns define the following decreasing order of solar wind exposure experience by the samples: $73261,8 \geq 73221,7 > 73281,6 > 73241,5 \approx 74240,16 \approx 74260,7 \gg 74220,84$. This sequence is also reflected in the He contents of the station 3 soils; however, the gray soils from station 4 appear to have anomalously high He abundances relative to C, N, and H. This probably reflects the greater amounts of ilmenite in the valley soils. Because of the chemical reactivity of solar wind C, N, and H ions, their retention in soils is expected to be relatively independent of soil mineralogy.

At all localities observed on the light mantle, a layer of light gray subsurface material occurs 5 to 10 cm below the medium gray surface soil (Schmitt). At station 3 accumulation of surface material was apparently interrupted by formation of Ballet Crater (Schmitt). Significantly, the solar wind exposure experienced by the light gray subsurface layer (73241,5) is about half that of the light gray portion of the marbled zone (73281,6). If these two samples have a common origin as has been suggested (Schmitt), the reason for the difference in exposure times is not clear. One explanation may be that the accumulation rate of this material was not constant, but, rather, slowed with time leading to a gradient decreasing with depth of solar wind exposure for soil grains. Alternatively, deposition of the light gray layer may have been accompanied by some grain size sorting which enriched the bottom layers in coarse particles and the upper ones in fine particles and led to apparently higher bulk concentrations of solar wind species in the upper layers. In either case, the impact event would have caused (a) inversion of strata, placing the layers of light gray soil least enriched in solar wind species at the very surface, and (b) marbling, the latter involving the soil layers more enriched in solar wind material which originally interfaced with the medium gray surface soil. Deposition of medium gray surface soil then resumed. This hypothesis for regolith evolution at station 3 can be tested when results from other analyses of the soils are reported and samples from drive tubes 73002 and 73001 become available for study. The similarity in solar wind exposure of 73221,7 and 73261,8 suggests that the gray surface soil may be accumulating at essentially the same rate as it did prior to the

ABUNDANCES OF C, N, H, He, AND S IN APOLLO 17 SOILS...

Chang, S. et al.

marbling event. Determination of the age of Ballet Crater should permit estimation of the accumulation rate of light mantle regolith.

The sulfur data in Table 2 conform with the simple mixing model proposed by Gibson and Moore:³ soils from station 3 derived predominantly from sulfur-poor massif material are low in S (< 700 ug/g), and soils from station 4 which contain a large sulfur-rich mare basalt component are high in S (> 1000 ug/g). Despite its mare basalt chemistry, the S content of orange soil is anomalously low and calls for a unique high temperature history. One very interesting result is that sample 73221,7 is very depleted in S relative to its companion soils. This observation suggests that either recent lateral mixing of soil on the light mantle with sulfur-rich foreign components is slow compared to regolith accumulation or the upper 0.5 cm of light mantle is derived in part from different source rocks than the underlying layers of soil. However, grain compositions (Apollo 17 LSIC)⁴ do not reveal gross mineralogic differences in these soils.

Simulation studies of solar wind related processes will also be reported. These include determination of implantation efficiencies of simulated solar wind C and N ions into bulk lunar fines and ilmenite targets. In Figure 1 is shown part of the gas release profile of 15013,3 fines which had been irradiated with $^{13}\text{CO}^+$ ions under conditions which approximate reimplantation of lunar atmospheric species. Evidently, the ^{13}C of the implanted species is evolved as ^{13}CO from 600–1150°C during vacuum pyrolysis. The release of ^{13}CO from unirradiated fines was indistinguishable from background levels, and release of $^{13}\text{CO}_2$ from the irradiated sample was not detectable above background. Although the ion energy is an order of magnitude too high, it is expected that CO derived from CO^+ implanted at lower energies would not exhibit a significantly different release pattern because of the reactivity of the carbon atoms formed upon impact. The coincidence in the release pattern of CO obtained from implanted CO^+ and C^+ (Chang et al.)⁵ suggests that a reimplanted atmospheric carbon component derived from the CO molecule ions cannot be distinguished by a characteristic pyrolysis temperature from C^+ which is either directly implanted by solar wind or formed by photodissociation of atmospheric CO and, then, reimplanted.

1. H. Sakai, I. R. Kaplan, C. Petrowski and J. W. Smith (1974) Analyses of C, N, S, He, H and metallic Fe in lunar samples by pyrolysis and hydrolysis. In preparation.
2. H. H. Schmitt (1973) Apollo 17 report on the valley of Taurus-Littrow, Science, 182, pp. 681-690.
3. E. K. Gibson, Jr. and G. W. Moore (1974) total sulfur abundances and distributions in the valley of Taurus-Littrow: evidence of mixing. In "Lunar Science V", The Lunar Science Institute, Houston.
4. Apollo 17 Lunar Sample Information Catalog (1973).
5. S. Chang, R. Mack and E. K. Gibson, Jr. (1973) Simulated solar wind implantation of carbon and nitrogen ions into terrestrial basalt and lunar fines. Proc. Fourth Lunar Sci. Conf., Vol. 2, pp. 1509-1522.

ABUNDANCES OF C, N, H, He, AND S IN APOLLO 17 SOILS...

Chang, S. et al.

Table 1. Abundances of Stable Light Elements Released by Vacuum Pyrolysis-combustion

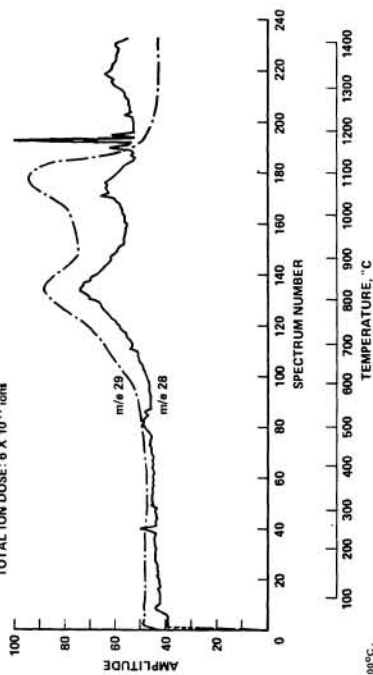
Sample (wt. %)	71221, 7 (0.184)	71241, 5 (1.040)	71281, 6 (0.949)	74220, 84 (1.245)	74240, 16 (1.488)	74260, 7 (1.246)
Gas Released	600° 1200° 1200°+O ₂	600° 1200° 1200°+O ₂	600° 1200° 1200°+O ₂	600° 1200° 1200°+O ₂	600° 1200° 1200°+O ₂	600° 1200° 1200°+O ₂
CH ₄ (μgC/g)	1	0.2	0.4	0	0.6	0.4
CO ₂ "	16	9.8	11.9	6.5	16	11.3
CO "	3	1.7	1.8	0.4	3.5	2.1
Total Carbon	81 μg/g	42 μg/g	81 μg/g	11 μg/g	50 μg/g	42 μg/g
N ₂ (μg/g)	44	21.9	39.8	7.3	16.4	13.1
He "	8	3.8	NM	2	19.4	24
H ₂ "	20	5.2	14.8	9.6	5.5	7.8

Figure 1.

IRRADIATED LUNAR FINES 16013.3

IMPLANTED ION: ¹³CO⁺

ION ENERGY: 29 KeV

ION FLUX: 2.7 X 10¹⁶ ions/hrTOTAL ION DOSE: 8 X 10¹⁷ ionsTable 2. Abundances of stable light elements released by acid hydrolysis in 6N H₂SO₄ at 120°C for >150 hr.

Sample	71221, 7	71241, 5	71281, 6	74220, 84	74240, 16	74260, 7
wt. (g)	0.111	0.958	0.951	0.992	1.142	0.931
CH ₄ (μgC/g)	7	3.5	7.1	3.4	0	1.3
CO ₂ "	28	15.8	16.2	10.8	5.8	16.5
CO "	6	1.2	2.0	5.8	0	1.8
H ₂ O (μgH/g)	110	606	677	578	564	1015
He (μg/g)	8	3.7	7.0	6.0	1.5	22
C ¹² (μg/g)	211	124	186	191	13.3	58
dFe (wt %)	0.51	0.51	0.45	0.49	0.035	0.18

a) Detection limits are as follows: CH₄, CO₂, CO, 0.2 μgC; N₂, 0.4 μgC; He, 0.08 μgC; H₂O, 0.5 μgC; H₂, 0.03 μgC. System blanks for gases are as follows: CH₄, 0.5 μgC; CO₂, 0.5 μgC; CO, 0.5 μgC; He, 0.03 μgC; H₂O, 0.5 μgC; H₂, 0.03 μgC. Prior to analysis all samples were outgassed under vacuum (410 Torr) at 150°C for 1 hr.

b) CH₄ represents indigenous methane plus methane derived from hydrolysis of carbonate-like carbon.

c) H₂ represents sum of solar wind hydrogen and hydrogen released by dissolution of metallic iron.

d) Metallic Fe is calculated from H₂ after correction for solar wind hydrogen from Table 1.

a) Significant measurements not performed at indicated temperature. In the case of CH₄, He and N₂ gases evolved at 600° pyrolysis were stored until after the 1200°C pyrolysis for measurement. For solar wind H₂, evolution is essentially complete at 600°C.