

**PINK ANGEL: ARGON AND XENON DIFFUSION, I-Xe CHRONOLOGY AND THE  $^{36}\text{Cl}$  PROBLEM.** G. Turner<sup>1</sup>, S.A. Crowther<sup>1</sup>, R. Burgess<sup>1</sup>, G.J. Wasserburg<sup>2</sup>, S.P. Kelley<sup>3</sup> and J.D. Gilmour<sup>1</sup> <sup>1</sup>SEAES, University of Manchester, Manchester M13 9PI, UK. [grenville.turner@manchester.ac.uk](mailto:grenville.turner@manchester.ac.uk). <sup>2</sup>California Institute of Technology, USA. <sup>3</sup>Open University, UK.

**Introduction:** The reported presence in Allende Pink Angel sodalite of excess  $^{36}\text{S}$  [1] attributed to  $^{36}\text{Cl}$  decay is at odds with the apparent absence of a corresponding excess of  $^{36}\text{Ar}$ , which is the major decay product of  $^{36}\text{Cl}$ . In an attempt to throw light on the problem we have reviewed new and existing data on the diffusion of Ar and Xe in sodalite, carried out new high resolution I-Xe analyses of Pink Angel, and devised a new methodology for searching for small excesses of  $^{36}\text{Ar}$  from  $^{36}\text{Cl}$  decay.

**Diffusion of Ar and Xe in sodalite:** Our experiments on neutron irradiated terrestrial sodalite indicate high retentivity of Cl-derived  $^{38}\text{Ar}$ . During stepped heating, maximum release occurred around 1100°C and release was essentially complete by 1250°C. An activation energy of 280 kJ/mol/K was calculated. Published data [2] from an I-Xe analysis of Pink Angel sodalite indicates peak release of I-derived  $^{128}\text{Xe}$  around 1200°C with significant release up to 1400°C. An activation energy for Xe diffusion of 460 kJ/mol/K is inferred. To calculate the implications for possible Ar and Xe loss over time scales appropriate to the early solar system we scale time,  $t$ , and absolute temperature,  $T$ , using the expression:  $1/T_2 = 1/T_1 + R/E \cdot \ln(t_2/t_1)$  where subscripts 1 and 2 refer to the laboratory and early solar system times and temperatures, respectively. To release essentially all of the Cl-correlated Ar and I-correlated Xe on a timescale of 1 Myr would require sustained temperatures of around 460°C and 740°C respectively.

**I-Xe analyses:** We have carried out laser stepped heating I-Xe analyses of Pink Angel sodalite using the RELAX resonance ionization spectrometer. Two samples, weighing 40ug and 80ug have been analysed with a total of 90 individual extractions. A well defined plateau indicates  $^{129}\text{I}/^{127}\text{I} = (0.93 \pm 0.02) \times 10^{-4}$ , corresponding to an I-Xe age  $3.2 \pm 0.3$  Ma after our monitor, the Shallowater achondrite. Lower  $^{129}\text{I}/^{127}\text{I}$  ratios in the early release could imply a 12% later loss of  $^{129}\text{Xe}$  concomitant with significant loss of  $^{36}\text{Ar}$ .

**$^{36}\text{Cl}$  Searches based on  $^{36}\text{Ar}$ :** We suggest a new method to search for  $^{36}\text{Ar}$  excesses which makes use of a plot of  $^{36}\text{Ar}/^{38}\text{Ar}$  vs.  $\text{Ca}/^{38}\text{Ar}$ . End members are trapped, cosmogenic and cosmogenic secondary neutron-induced argon. In suitable circumstances excess  $^{36}\text{Ar}$  would plot above this mixing triangle. Determination of Ca using a reactor irradiation requires low fluences and Cd-shielding to minimise  $^{38}\text{Ar}$  production from Cl.

In conclusion we note that absence of excess  $^{36}\text{Ar}$  in spite of the high retentivity of sodalite leaves open the possibility that either the reported  $^{36}\text{S}$  excess is an artifact or else it is inherited from an unidentified pre-existing Cl-rich phase which has lost  $^{36}\text{Ar}$  and the apparent correlation with Cl/S represents a two component mixing line and not an isochron.

**References:** [1] Hsu W. et al. 2006. *Astrophysical Journal* 640:525-529. [2] Swindle T.D. et al. 1988. *Geochimica Cosmochimica Acta* 52:2215-2227.